

2-Methyl-3-(2-methylphenyl)-7-nitroquinazolin-4(3H)-one

Adel S. El-Azab,^{a,b,†} Alaa A.-M. Abdel-Aziz,^{a,c} Seik Weng Ng^{d,e} and Edward R. T. Tiekink^{d*}

^aDepartment of Pharmaceutical Chemistry, College of Pharmacy, King Saud University, Riyadh 11451, Saudi Arabia, ^bDepartment of Organic Chemistry, Faculty of Pharmacy, Al-Azhar University, Cairo 11884, Egypt, ^cDepartment of Medicinal Chemistry, Faculty of Pharmacy, University of Mansoura, Mansoura 35516, Egypt, ^dDepartment of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia, and ^eChemistry Department, Faculty of Science, King Abdulaziz University, PO Box 80203 Jeddah, Saudi Arabia

Correspondence e-mail: edward.tiekink@gmail.com

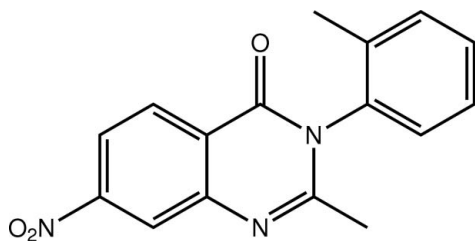
Received 18 February 2012; accepted 18 February 2012

Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.035; wR factor = 0.099; data-to-parameter ratio = 13.8.

In the title methaqualone analogue, $\text{C}_{16}\text{H}_{13}\text{N}_3\text{O}_3$, the 2-tolyl group is almost orthogonal [dihedral angle = $85.20(5)^\circ$] to the fused ring system (r.m.s. deviation of fitted non-H atoms = 0.029 Å). In the crystal, twofold symmetry generates two-molecule aggregates linked by $\text{C}-\text{H}\cdots\text{O}$ and $\pi-\pi$ interactions [ring centroid-centroid distance = $3.4967(6)$ Å].

Related literature

For recent studies on synthesis, drug discovery and crystal structures of quinazoline-4(3H)-one derivatives, see: El-Azab *et al.* (2010, 2012). For the anti-convulsant activity of the title methaqualone analogue, see: El-Azab *et al.* (2011). For a related structure, see: Stephenson *et al.* (2011).



Experimental

Crystal data

$\text{C}_{16}\text{H}_{13}\text{N}_3\text{O}_3$
 $M_r = 295.29$
 Monoclinic, $C2/c$
 $a = 14.4614(5)$ Å

$b = 16.5383(4)$ Å
 $c = 12.9968(4)$ Å
 $\beta = 119.072(4)^\circ$
 $V = 2716.78(14)$ Å³

$Z = 8$
 Cu $K\alpha$ radiation
 $\mu = 0.85$ mm⁻¹

$T = 100$ K
 $0.25 \times 0.25 \times 0.25$ mm

Data collection

Agilent SuperNova Dual diffractometer with an Atlas detector
 Absorption correction: multi-scan (*CrysAlis PRO*; Agilent, 2011)
 $T_{\min} = 0.967$, $T_{\max} = 0.998$

5450 measured reflections
 2783 independent reflections
 2570 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.013$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.035$
 $wR(F^2) = 0.099$
 $S = 1.02$
 2783 reflections

201 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.24$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.24$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C11}-\text{H11}\cdots\text{O1}^i$	0.95	2.54	3.4530 (15)	161

Symmetry code: (i) $-x + 1, y, -z + \frac{1}{2}$.

Data collection: *CrysAlis PRO* (Agilent, 2011); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997) and *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *publCIF* (Westrip, 2010).

This work was supported by the Research Center of Pharmacy, King Saud University, Riyadh, Saudi Arabia. The authors also thank the Ministry of Higher Education (Malaysia) for funding structural studies through the High-Impact Research scheme (grant No. UM-C/HIR/MOHE/SC/12).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6642).

References

- Agilent (2011). *CrysAlis PRO*. Agilent Technologies, Yarnton, Oxfordshire, England.
 Brandenburg, K. (2006). *DIAMOND*. Crystal Impact GbR, Bonn, Germany.
 El-Azab, A. S., Al-Omar, M. A., Abdel-Aziz, A. A.-M., Abdel-Aziz, N. I., El-Sayed, M. A.-A., Aleisa, A. M., Sayed-Ahmed, M. M. & Abdel-Hamide, S. G. (2010). *Eur. J. Med. Chem.* **45**, 4188–4198.
 El-Azab, A. S. & ElTahir, K. H. (2012). *Bioorg. Med. Chem. Lett.* **22**, 327–333.
 El-Azab, A. S., ElTahir, K. H. & Attia, S. M. (2011). *Monatsh. Chem.* **142**, 837–848.
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
 Stephenson, K. A., Wilson, A. A., Houle, S. & Vasdev, N. (2011). *Bioorg. Med. Chem. Lett.* **21**, 5506–5509.
 Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.

† Additional correspondence author, e-mail: adelazaba@yahoo.com.

supplementary materials

Acta Cryst. (2012). E68, o863 [doi:10.1107/S1600536812007350]

2-Methyl-3-(2-methylphenyl)-7-nitroquinazolin-4(3H)-one

Adel S. El-Azab, Alaa A.-M. Abdel-Aziz, Seik Weng Ng and Edward R. T. Tiekink

Comment

The title methaqualone analogue, 2-methyl-7-nitro-3-*o*-tolylquinazolin-4(3H)-one (I), has been investigated previously for its anti-convulsant activity (El-Azab *et al.*, 2012) as quinazoline-4(3H)-one derivatives are known for their various biological activities (El-Azab *et al.*, 2011, 2010). Herein, the crystal and molecular structure of (I) is described. A related structure with a similar conformation has been reported recently (Stephenson *et al.*, 2011).

The ten atoms comprising the fused ring system in (I), Fig. 1, are almost co-planar with the r.m.s. deviation for the non-hydrogen atoms being 0.029 Å. The 2-tolyl group is approximately orthogonal to this plane, forming a dihedral angle of 85.20 (5)°. The nitro group is almost co-planar with the ring to which it is connected as seen in the value of the O2—N3—C3—C2 torsion angle of 179.07 (10)°.

The main feature of the crystal packing is the formation of two molecule aggregates sustained by C—H···O, Table 1, and π – π interactions, Fig. 2. The π – π interactions occur between (C1–C6) rings, *i.e.* the C₆ group within the fused ring system, with a ring centroid-to-centroid distance of 3.4967 (6) Å [symmetry operation: 1 - *x*, *y*, 1/2 - *z*]. There are no specific intermolecular interactions between the two-molecule aggregates [generated by 2-fold symmetry]. They assemble into layers in the *bc* plane, Fig. 3, and these stack along the *a* axis, Fig. 4.

Experimental

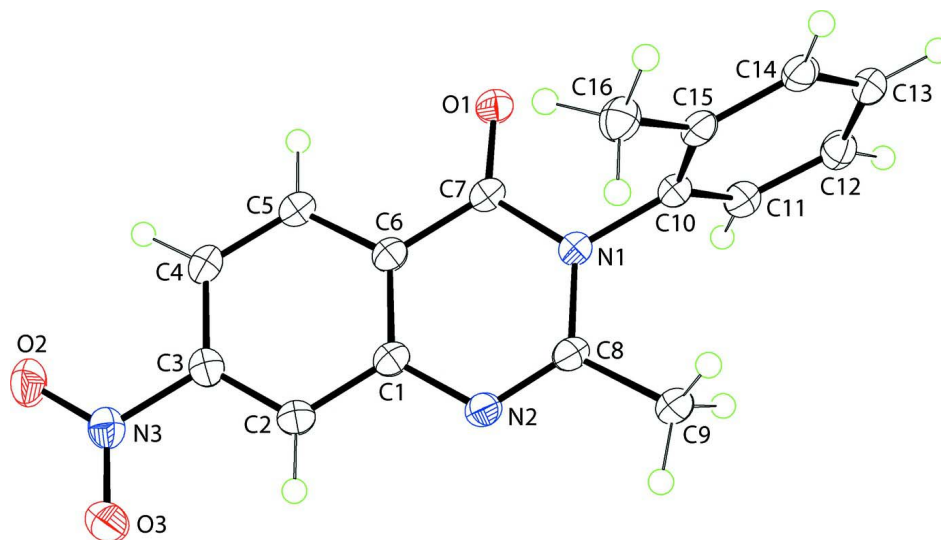
4-Nitroanthranilic acid (1.82 g, 10 mmol) was refluxed with acetic anhydride (20 ml) for 3 h. The reaction mixture was cooled, filtered, washed with petroleum ether, and dried to yield 2-methyl-7-nitro-4H-3,1-benzoxazin-4-one as a solid compound. This was refluxed with *o*-toluidine (1.18 g, 11 mmol) in pyridine (30 ml) for 8 h. The reaction mixture was cooled, the solvent was removed under reduced pressure and the residue was triturated with water and filtered. The solid obtained was dried, chromatographed with CHCl₃ and recrystallized from its EtOH solution as colourless cubes.

Refinement

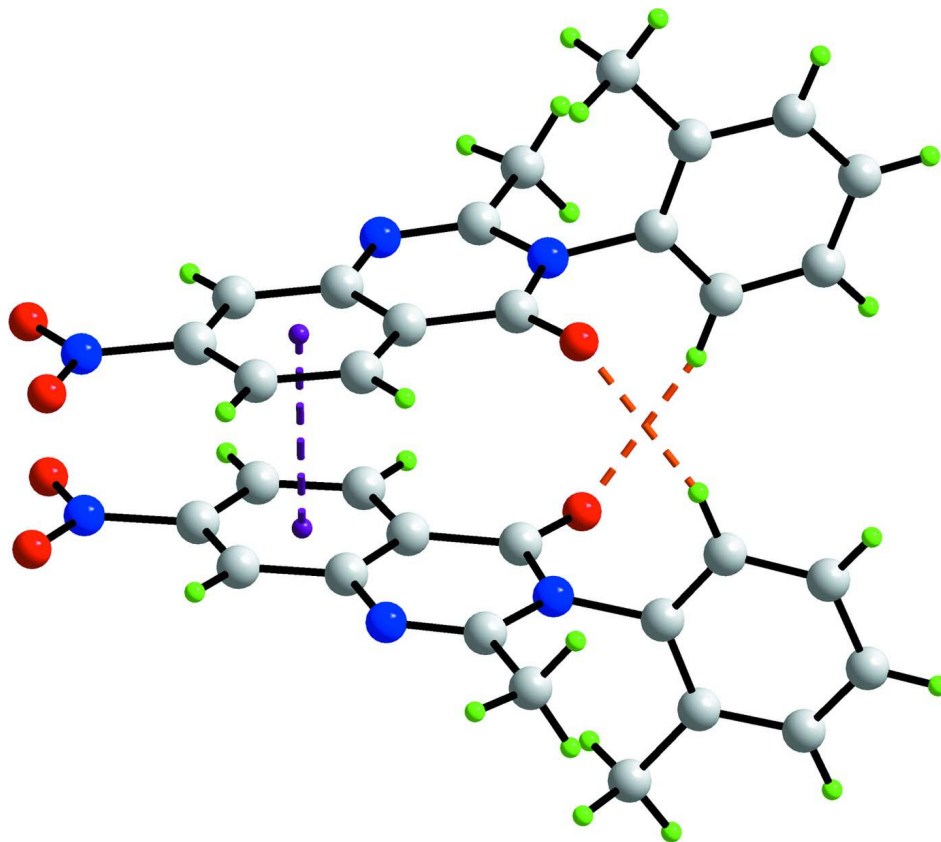
Carbon-bound H-atoms were placed in calculated positions [C—H = 0.95 to 0.98 Å, $U_{\text{iso}}(\text{H}) = 1.2$ to $1.5U_{\text{eq}}(\text{C})$] and were included in the refinement in the riding model approximation.

Computing details

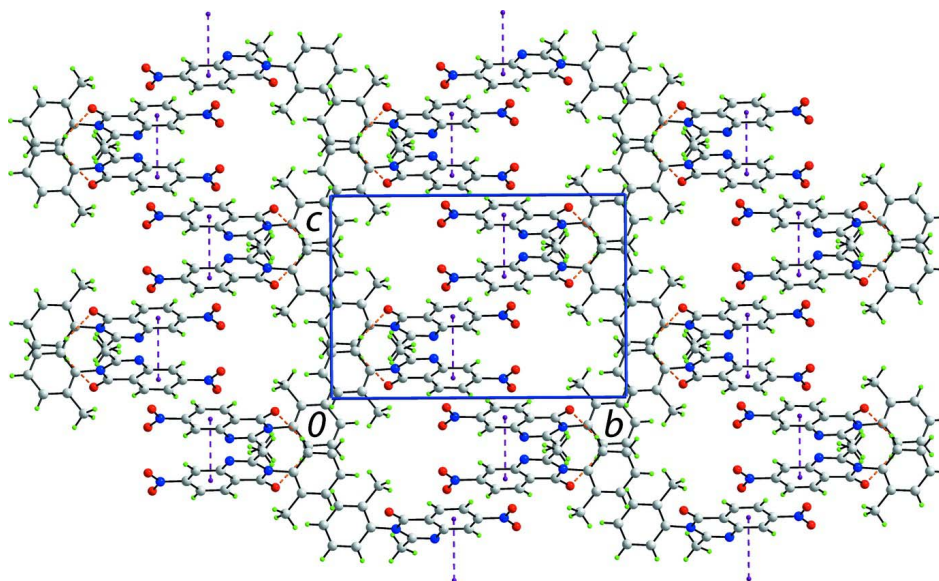
Data collection: *CrysAlis PRO* (Agilent, 2011); cell refinement: *CrysAlis PRO* (Agilent, 2011); data reduction: *CrysAlis PRO* (Agilent, 2011); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997) and *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *publCIF* (Westrip, 2010).

**Figure 1**

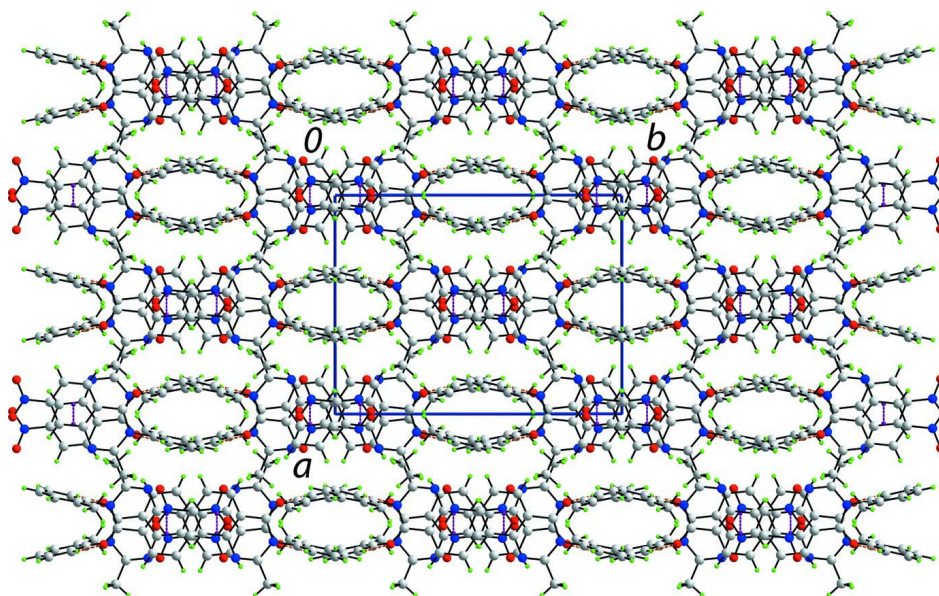
The molecular structure of (I) showing displacement ellipsoids at the 50% probability level.

**Figure 2**

A view of the two-molecular aggregate in (I) sustained by C—H...O and π - π interactions shown as orange and purple dashed lines, respectively.


Figure 3

A view of the assembly of two-molecule aggregates in the bc plane in (I). The C—H \cdots O and π — π interactions are shown as orange and purple dashed lines, respectively.


Figure 4

A view in projection down the a axis of the unit-cell contents of (I). The C—H \cdots O and π — π interactions are shown as orange and purple dashed lines, respectively.

2-Methyl-3-(2-methylphenyl)-7-nitroquinazolin-4(3H)-one

Crystal data

$C_{16}H_{13}N_3O_3$

$M_r = 295.29$

Monoclinic, $C2/c$

Hall symbol: $-C 2yc$

$a = 14.4614 (5) \text{ \AA}$

$b = 16.5383 (4) \text{ \AA}$

$c = 12.9968 (4) \text{ \AA}$

$\beta = 119.072 (4)^\circ$

$V = 2716.78 (14) \text{ \AA}^3$
 $Z = 8$
 $F(000) = 1232$
 $D_x = 1.444 \text{ Mg m}^{-3}$
 Cu $K\alpha$ radiation, $\lambda = 1.54184 \text{ \AA}$
 Cell parameters from 3546 reflections

$\theta = 3.9\text{--}76.4^\circ$
 $\mu = 0.85 \text{ mm}^{-1}$
 $T = 100 \text{ K}$
 Cube, colourless
 $0.25 \times 0.25 \times 0.25 \text{ mm}$

Data collection

Agilent SuperNova Dual
 diffractometer with an Atlas detector
 Radiation source: SuperNova (Cu) X-ray
 Source
 Mirror monochromator
 Detector resolution: $10.4041 \text{ pixels mm}^{-1}$
 ω scan
 Absorption correction: multi-scan
 (*CrysAlis PRO*; Agilent, 2011)

$T_{\min} = 0.967, T_{\max} = 0.998$
 5450 measured reflections
 2783 independent reflections
 2570 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.013$
 $\theta_{\max} = 76.6^\circ, \theta_{\min} = 4.4^\circ$
 $h = -18 \rightarrow 17$
 $k = -20 \rightarrow 18$
 $l = -10 \rightarrow 16$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.035$
 $wR(F^2) = 0.099$
 $S = 1.02$
 2783 reflections
 201 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0566P)^2 + 1.6285P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.24 \text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.24 \text{ e \AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.59490 (6)	0.19268 (5)	0.43782 (7)	0.0250 (2)
O2	0.65765 (7)	0.61062 (5)	0.45232 (9)	0.0338 (2)
O3	0.48735 (7)	0.62439 (5)	0.36278 (8)	0.0330 (2)
N1	0.41702 (7)	0.21605 (6)	0.35394 (8)	0.0201 (2)
N2	0.34237 (7)	0.34733 (6)	0.30533 (8)	0.0217 (2)
N3	0.56777 (8)	0.58294 (6)	0.40625 (9)	0.0247 (2)
C1	0.44368 (9)	0.37812 (7)	0.35225 (9)	0.0196 (2)
C2	0.45504 (9)	0.46277 (7)	0.35369 (9)	0.0208 (2)
H2	0.3949	0.4972	0.3210	0.025*
C3	0.55556 (9)	0.49398 (7)	0.40369 (10)	0.0214 (2)
C4	0.64664 (9)	0.44647 (7)	0.45173 (10)	0.0225 (2)

H4	0.7148	0.4707	0.4859	0.027*
C5	0.63564 (9)	0.36354 (7)	0.44855 (9)	0.0215 (2)
H5	0.6964	0.3298	0.4795	0.026*
C6	0.53454 (9)	0.32924 (7)	0.39953 (9)	0.0195 (2)
C7	0.52197 (8)	0.24124 (7)	0.39980 (9)	0.0200 (2)
C8	0.33210 (8)	0.26981 (7)	0.30874 (9)	0.0211 (2)
C9	0.22363 (9)	0.23499 (7)	0.26233 (11)	0.0273 (3)
H9A	0.1715	0.2788	0.2356	0.041*
H9B	0.2199	0.2044	0.3248	0.041*
H9C	0.2084	0.1989	0.1962	0.041*
C10	0.39793 (8)	0.13076 (7)	0.36267 (10)	0.0207 (2)
C11	0.37329 (9)	0.08065 (7)	0.26703 (10)	0.0236 (2)
H11	0.3732	0.1011	0.1987	0.028*
C12	0.34882 (9)	0.00019 (7)	0.27286 (11)	0.0252 (3)
H12	0.3291	-0.0345	0.2071	0.030*
C13	0.35326 (9)	-0.02949 (7)	0.37522 (11)	0.0242 (3)
H13	0.3371	-0.0846	0.3798	0.029*
C14	0.38125 (9)	0.02132 (7)	0.47070 (10)	0.0232 (2)
H14	0.3863	-0.0001	0.5411	0.028*
C15	0.40209 (8)	0.10313 (7)	0.46597 (10)	0.0214 (2)
C16	0.42659 (10)	0.15867 (8)	0.56764 (10)	0.0270 (3)
H16A	0.4905	0.1899	0.5866	0.040*
H16B	0.4381	0.1266	0.6363	0.040*
H16C	0.3671	0.1957	0.5463	0.040*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0190 (4)	0.0239 (4)	0.0292 (4)	0.0027 (3)	0.0094 (3)	0.0001 (3)
O2	0.0311 (5)	0.0281 (5)	0.0405 (5)	-0.0096 (4)	0.0161 (4)	-0.0045 (4)
O3	0.0324 (5)	0.0247 (4)	0.0405 (5)	0.0026 (4)	0.0166 (4)	0.0039 (4)
N1	0.0172 (5)	0.0203 (5)	0.0208 (5)	-0.0004 (3)	0.0077 (4)	0.0000 (3)
N2	0.0180 (4)	0.0233 (5)	0.0217 (5)	0.0007 (4)	0.0080 (4)	0.0022 (4)
N3	0.0284 (5)	0.0241 (5)	0.0238 (5)	-0.0033 (4)	0.0143 (4)	-0.0008 (4)
C1	0.0190 (5)	0.0242 (5)	0.0158 (5)	0.0000 (4)	0.0087 (4)	0.0003 (4)
C2	0.0213 (5)	0.0237 (6)	0.0182 (5)	0.0017 (4)	0.0101 (4)	0.0015 (4)
C3	0.0265 (6)	0.0218 (5)	0.0183 (5)	-0.0021 (4)	0.0126 (4)	-0.0007 (4)
C4	0.0195 (5)	0.0275 (6)	0.0194 (5)	-0.0046 (4)	0.0088 (4)	-0.0023 (4)
C5	0.0185 (5)	0.0262 (6)	0.0187 (5)	0.0003 (4)	0.0082 (4)	-0.0005 (4)
C6	0.0194 (5)	0.0230 (5)	0.0160 (5)	-0.0002 (4)	0.0087 (4)	-0.0005 (4)
C7	0.0170 (5)	0.0245 (6)	0.0173 (5)	0.0003 (4)	0.0075 (4)	-0.0010 (4)
C8	0.0171 (5)	0.0251 (5)	0.0188 (5)	0.0004 (4)	0.0069 (4)	0.0014 (4)
C9	0.0175 (5)	0.0265 (6)	0.0326 (6)	-0.0001 (4)	0.0080 (5)	0.0038 (5)
C10	0.0161 (5)	0.0208 (5)	0.0226 (5)	-0.0002 (4)	0.0073 (4)	0.0006 (4)
C11	0.0216 (5)	0.0262 (6)	0.0217 (5)	-0.0001 (4)	0.0094 (4)	0.0001 (4)
C12	0.0223 (6)	0.0252 (6)	0.0260 (6)	-0.0021 (4)	0.0101 (5)	-0.0049 (4)
C13	0.0182 (5)	0.0221 (5)	0.0291 (6)	-0.0014 (4)	0.0090 (5)	0.0007 (4)
C14	0.0164 (5)	0.0269 (6)	0.0240 (6)	-0.0010 (4)	0.0081 (4)	0.0030 (4)
C15	0.0145 (5)	0.0252 (6)	0.0219 (5)	-0.0002 (4)	0.0068 (4)	-0.0010 (4)
C16	0.0270 (6)	0.0301 (6)	0.0227 (6)	-0.0035 (5)	0.0112 (5)	-0.0025 (5)

Geometric parameters (Å, °)

O1—C7	1.2227 (14)	C8—C9	1.4955 (15)
O2—N3	1.2253 (13)	C9—H9A	0.9800
O3—N3	1.2261 (14)	C9—H9B	0.9800
N1—C8	1.3938 (14)	C9—H9C	0.9800
N1—C7	1.3976 (14)	C10—C11	1.3887 (16)
N1—C10	1.4522 (14)	C10—C15	1.3915 (16)
N2—C8	1.2939 (15)	C11—C12	1.3885 (16)
N2—C1	1.3818 (14)	C11—H11	0.9500
N3—C3	1.4803 (14)	C12—C13	1.3899 (17)
C1—C6	1.4046 (15)	C12—H12	0.9500
C1—C2	1.4086 (15)	C13—C14	1.3863 (17)
C2—C3	1.3724 (16)	C13—H13	0.9500
C2—H2	0.9500	C14—C15	1.3939 (16)
C3—C4	1.3939 (16)	C14—H14	0.9500
C4—C5	1.3790 (16)	C15—C16	1.5035 (15)
C4—H4	0.9500	C16—H16A	0.9800
C5—C6	1.4000 (15)	C16—H16B	0.9800
C5—H5	0.9500	C16—H16C	0.9800
C6—C7	1.4670 (15)		
C8—N1—C7	122.78 (9)	C8—C9—H9A	109.5
C8—N1—C10	119.37 (9)	C8—C9—H9B	109.5
C7—N1—C10	117.68 (9)	H9A—C9—H9B	109.5
C8—N2—C1	117.70 (9)	C8—C9—H9C	109.5
O2—N3—O3	124.04 (10)	H9A—C9—H9C	109.5
O2—N3—C3	117.96 (10)	H9B—C9—H9C	109.5
O3—N3—C3	117.99 (10)	C11—C10—C15	122.65 (11)
N2—C1—C6	123.15 (10)	C11—C10—N1	119.05 (10)
N2—C1—C2	117.64 (10)	C15—C10—N1	118.27 (10)
C6—C1—C2	119.20 (10)	C12—C11—C10	118.98 (11)
C3—C2—C1	118.02 (10)	C12—C11—H11	120.5
C3—C2—H2	121.0	C10—C11—H11	120.5
C1—C2—H2	121.0	C11—C12—C13	119.74 (11)
C2—C3—C4	123.58 (10)	C11—C12—H12	120.1
C2—C3—N3	118.14 (10)	C13—C12—H12	120.1
C4—C3—N3	118.28 (10)	C14—C13—C12	120.02 (11)
C5—C4—C3	118.49 (10)	C14—C13—H13	120.0
C5—C4—H4	120.8	C12—C13—H13	120.0
C3—C4—H4	120.8	C13—C14—C15	121.61 (11)
C4—C5—C6	119.76 (10)	C13—C14—H14	119.2
C4—C5—H5	120.1	C15—C14—H14	119.2
C6—C5—H5	120.1	C10—C15—C14	116.90 (10)
C5—C6—C1	120.93 (10)	C10—C15—C16	121.90 (10)
C5—C6—C7	120.27 (10)	C14—C15—C16	121.19 (10)
C1—C6—C7	118.77 (10)	C15—C16—H16A	109.5
O1—C7—N1	121.42 (10)	C15—C16—H16B	109.5
O1—C7—C6	124.68 (10)	H16A—C16—H16B	109.5
N1—C7—C6	113.88 (9)	C15—C16—H16C	109.5

N2—C8—N1	123.66 (10)	H16A—C16—H16C	109.5
N2—C8—C9	118.89 (10)	H16B—C16—H16C	109.5
N1—C8—C9	117.45 (10)		
C8—N2—C1—C6	2.02 (16)	C1—C6—C7—O1	179.56 (10)
C8—N2—C1—C2	-176.98 (10)	C5—C6—C7—N1	176.62 (9)
N2—C1—C2—C3	177.87 (10)	C1—C6—C7—N1	-1.68 (14)
C6—C1—C2—C3	-1.17 (15)	C1—N2—C8—N1	-1.83 (16)
C1—C2—C3—C4	0.77 (17)	C1—N2—C8—C9	177.94 (10)
C1—C2—C3—N3	-179.48 (9)	C7—N1—C8—N2	-0.21 (17)
O2—N3—C3—C2	179.07 (10)	C10—N1—C8—N2	174.96 (10)
O3—N3—C3—C2	-0.50 (15)	C7—N1—C8—C9	-179.98 (10)
O2—N3—C3—C4	-1.17 (15)	C10—N1—C8—C9	-4.82 (15)
O3—N3—C3—C4	179.26 (10)	C8—N1—C10—C11	88.12 (13)
C2—C3—C4—C5	0.36 (17)	C7—N1—C10—C11	-96.47 (12)
N3—C3—C4—C5	-179.39 (9)	C8—N1—C10—C15	-90.00 (12)
C3—C4—C5—C6	-1.07 (16)	C7—N1—C10—C15	85.41 (13)
C4—C5—C6—C1	0.66 (16)	C15—C10—C11—C12	1.87 (17)
C4—C5—C6—C7	-177.60 (10)	N1—C10—C11—C12	-176.16 (10)
N2—C1—C6—C5	-178.50 (10)	C10—C11—C12—C13	-2.42 (17)
C2—C1—C6—C5	0.48 (16)	C11—C12—C13—C14	0.49 (17)
N2—C1—C6—C7	-0.22 (15)	C12—C13—C14—C15	2.14 (17)
C2—C1—C6—C7	178.77 (9)	C11—C10—C15—C14	0.65 (16)
C8—N1—C7—O1	-179.25 (10)	N1—C10—C15—C14	178.70 (9)
C10—N1—C7—O1	5.51 (15)	C11—C10—C15—C16	-178.64 (10)
C8—N1—C7—C6	1.94 (14)	N1—C10—C15—C16	-0.59 (16)
C10—N1—C7—C6	-173.30 (9)	C13—C14—C15—C10	-2.66 (16)
C5—C6—C7—O1	-2.15 (17)	C13—C14—C15—C16	176.63 (10)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C11—H11 \cdots O1 ⁱ	0.95	2.54	3.4530 (15)	161

Symmetry code: (i) $-x+1, y, -z+1/2$.